

# Innovative Perovskite Materials for High-performing Reversible Solid Oxide Cells

## BACKGROUND OF THE STUDY

### Green-ticks on Reversible Solid Oxide Cell (RSOC)

- ✓ Combines the capability of electrolytic cell and fuel cell into one device;
- ✓ Convert excess generated renewable energy into green hydrogen for storage;
- ✓ Enables on-demand production of green hydrogen and electricity;
- ✓ Can facilitate a just net zero society with hydrogen as the main energy carrier.

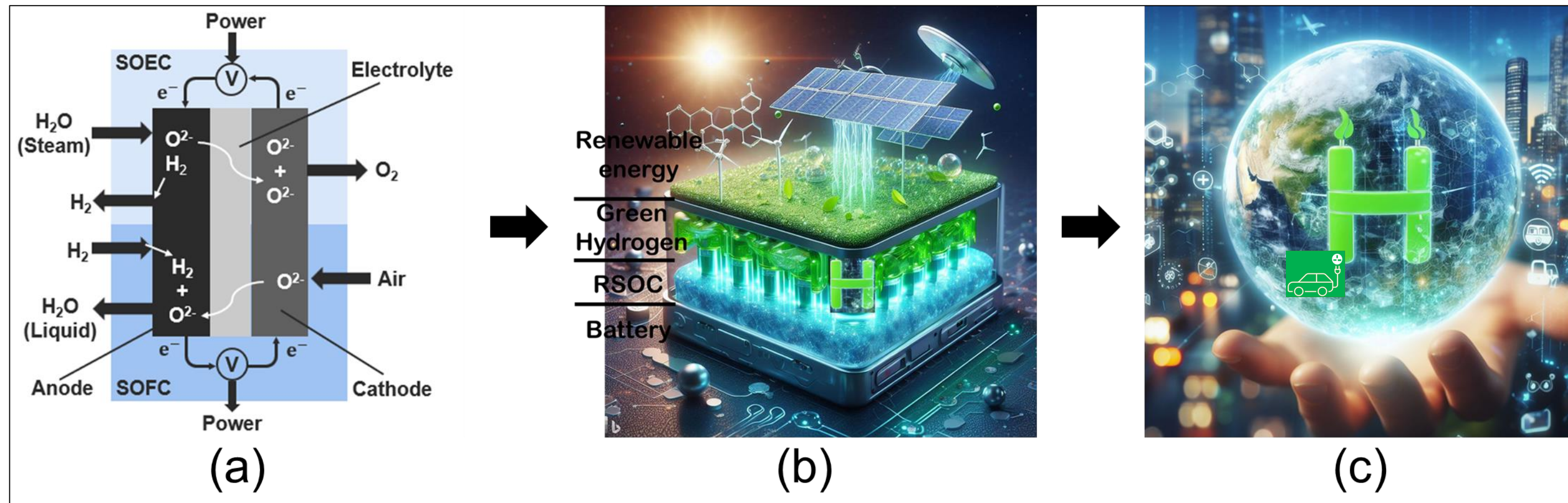


Figure 1: (a) a RSOC, (b) a net zero energy system built on the potentials of renewable energy and green hydrogen, and (c) a net zero society with hydrogen as the primary energy carrier.

### Drawback to Full Commercialisation of RSOC

- Stringent electrochemical requirements: high ionic and electronic conductivity, high catalytic activity, durability, stability, etc;
- High operating temperature requirement;
- Need for innovative electrode materials.

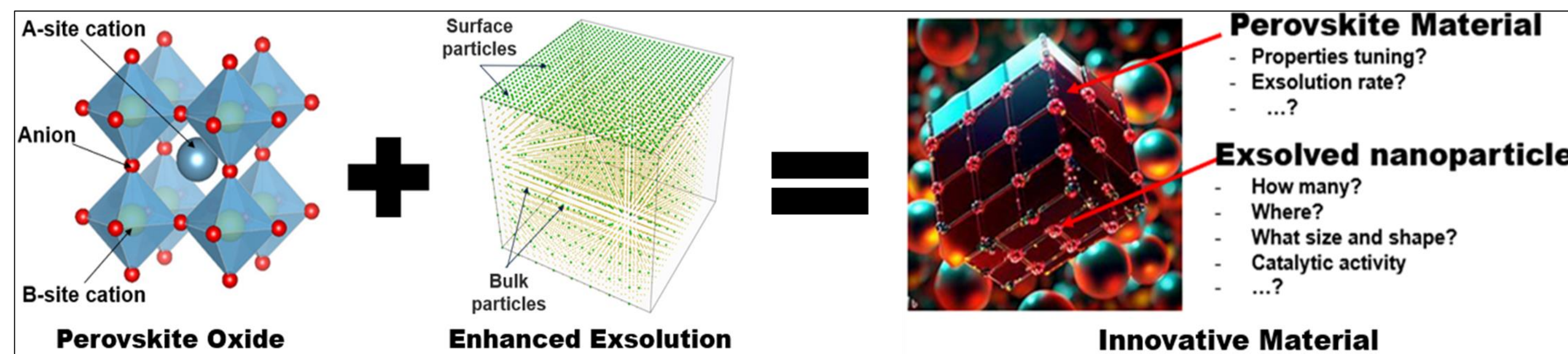


Figure 2: Schematics of innovative perovskite material enabled by enhanced exsolution process

## OBJECTIVES OF THE RESEARCH

This research aims to develop novel perovskite materials capable of surface and bulk exsolution processes (in-situ formation of metallic nanoparticles) to fulfil the multiple electrochemical requirements of RSOCs.

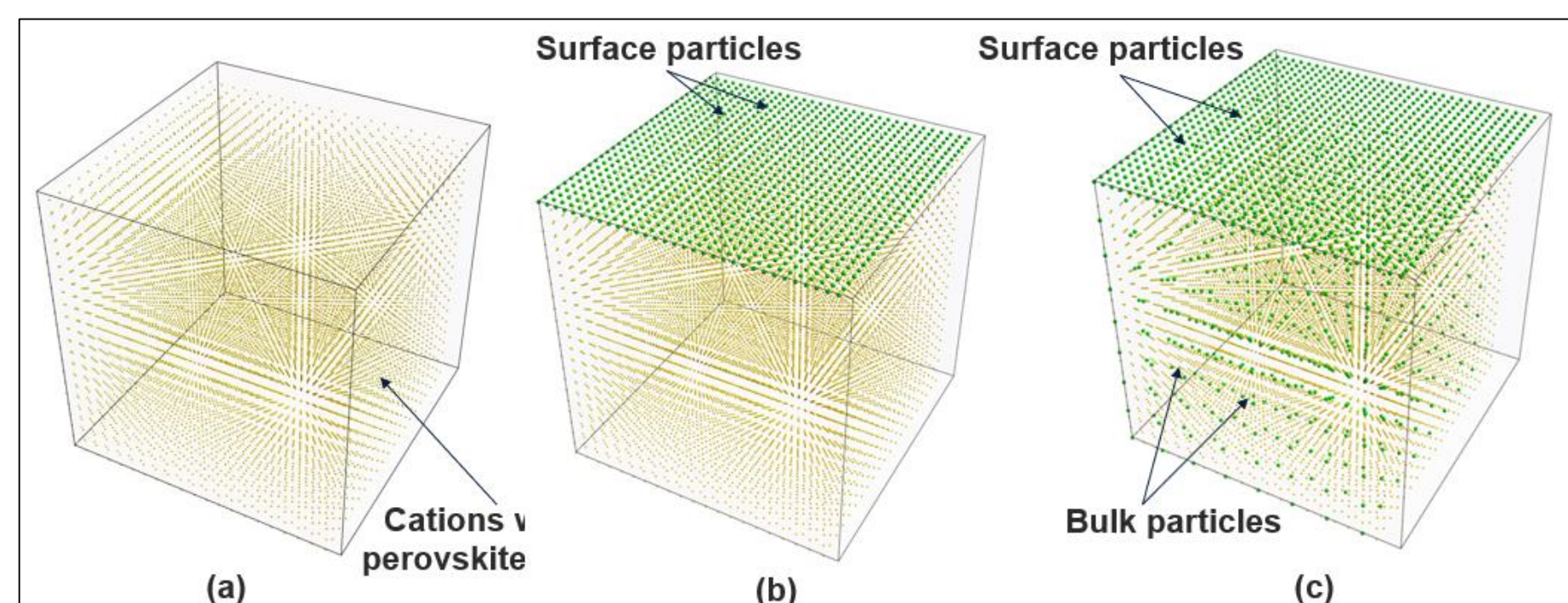
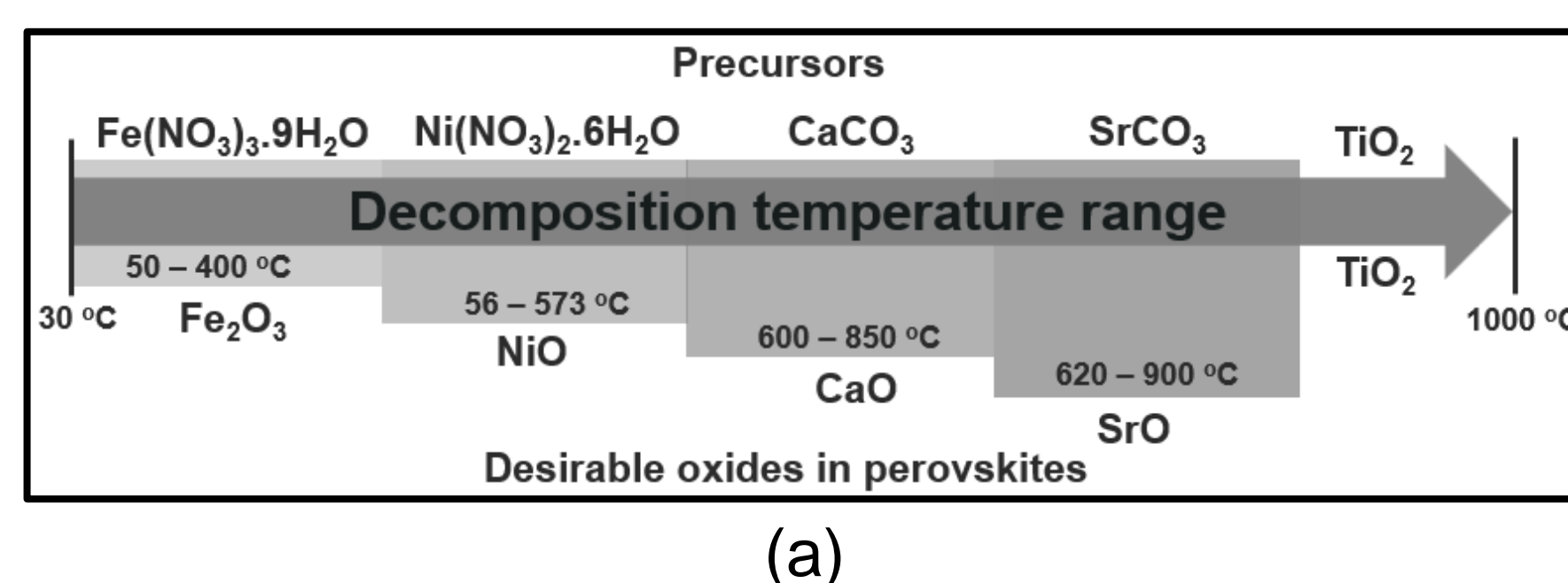
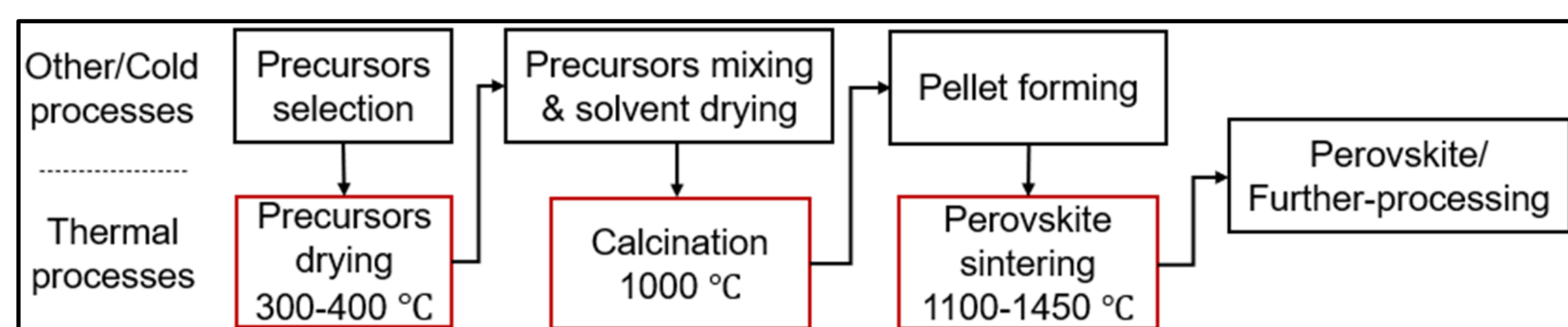


Figure 3: Schematics of a nanoscale representation of a perovskite lattice at (a) cations segregation, (b) surface exsolution, and (c) bulk and surface exsolution.

## METHODOLOGY



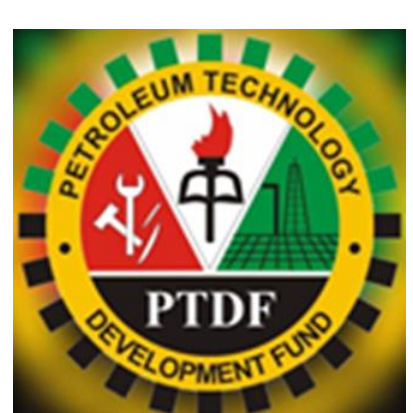
(a)



(b)

Figure 4: (a) Predicted calcination temperature of the perovskite through a thermogravimetry analysis (b) Processing steps in the solid-state synthesis method for the perovskite material.

## ACKNOWLEDGEMENTS



Research Funder



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## RESULTS

### Perovskites synthesized

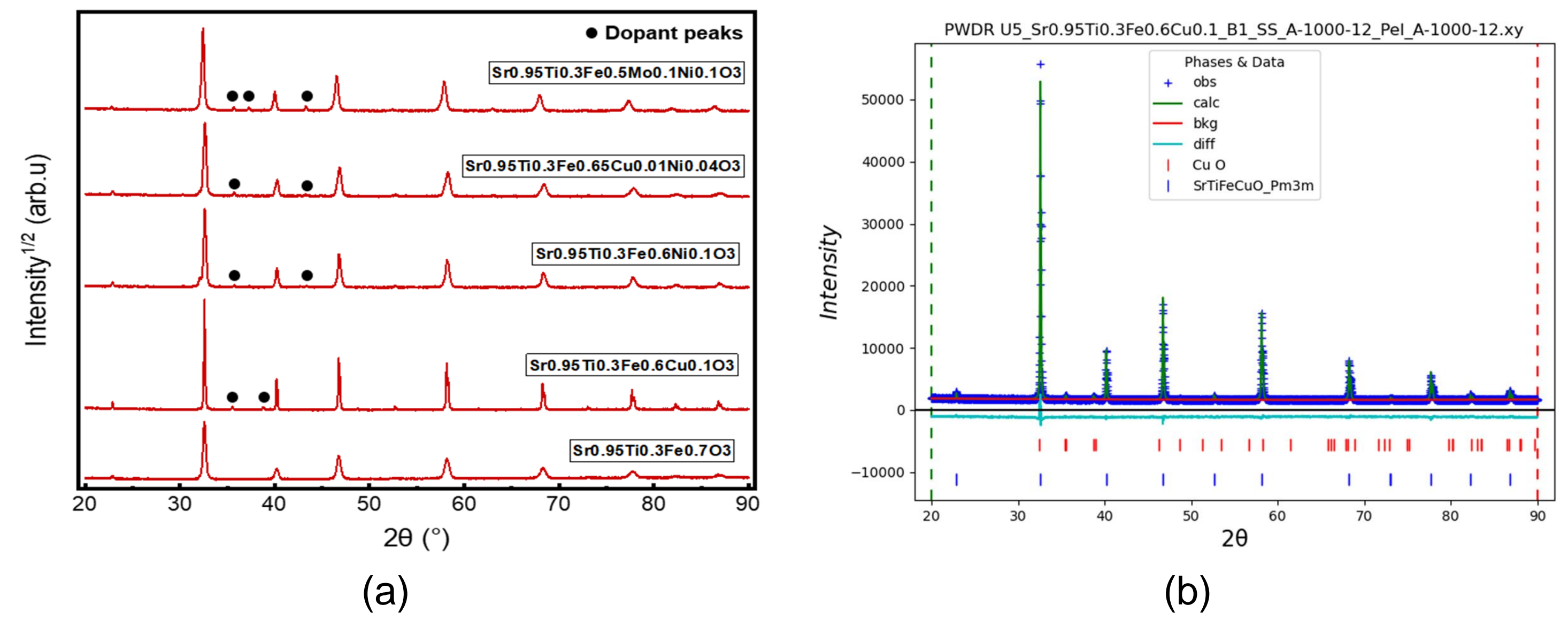


Figure 5: (a) Room temperature powder XRD pattern of four perovskites synthesized from doping A-site deficient  $\text{Sr}_{0.95}\text{Ti}_{0.3}\text{Fe}_{0.7}\text{O}_3$  with selected dopants. (a) Refined peaks of  $\text{Sr}_{0.95}\text{Ti}_{0.3}\text{Fe}_{0.6}\text{Cu}_{0.1}\text{O}_3$

### Crystal Structure of the Perovskites from Rietveld Refinement

Perovskite	Cell Parameter a (Å)	Cell Type	Cell Volume (Å <sup>3</sup> )
$\text{Sr}_{0.95}\text{Ti}_{0.3}\text{Fe}_{0.7}\text{O}_3$	3.8871	Cubic	58.7323
$\text{Sr}_{0.95}\text{Ti}_{0.3}\text{Fe}_{0.7}\text{Cu}_{0.1}\text{O}_3$	3.8822	Cubic	58.5088

### Exsolution on the Reduced Perovskites

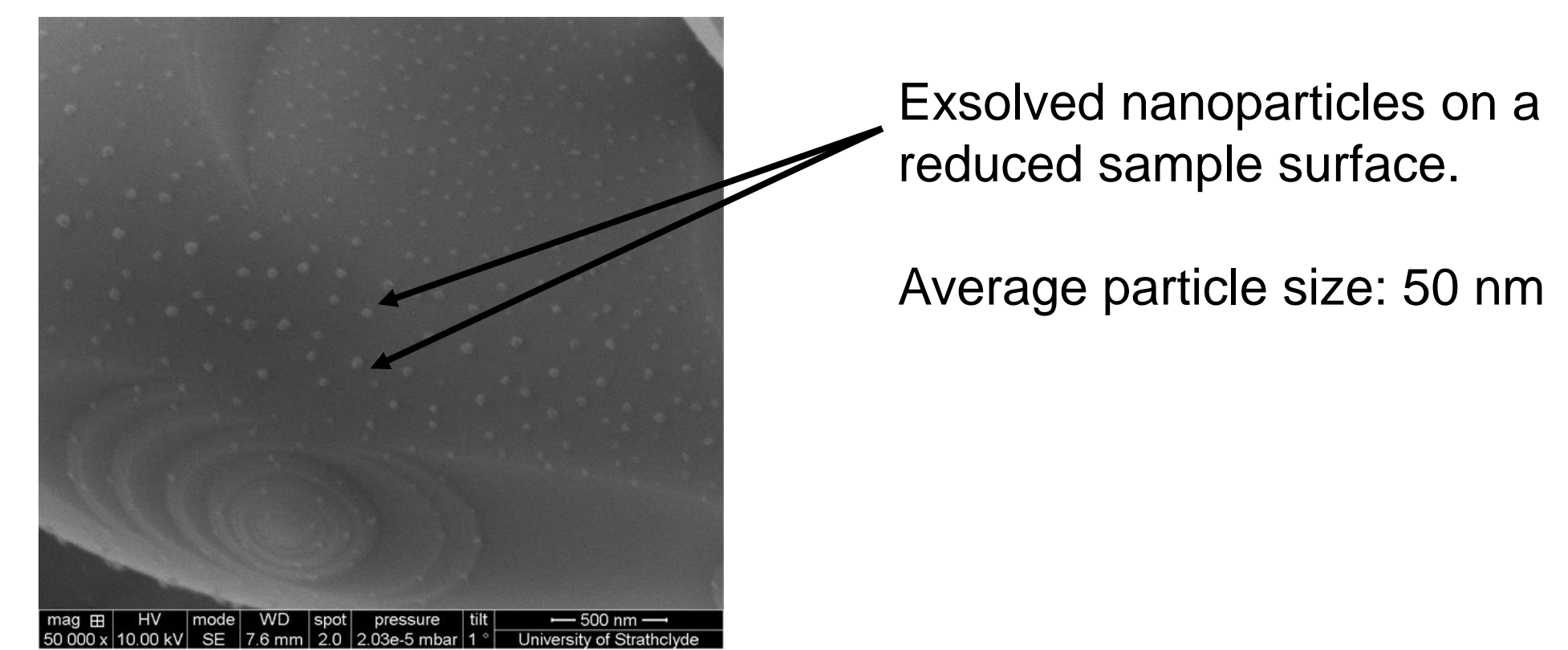


Figure 6: SEM image of  $\text{Sr}_{0.95}\text{Ti}_{0.3}\text{Fe}_{0.5}\text{Mo}_{0.1}\text{Ni}_{0.1}\text{O}_3$  reduced in a 5%  $\text{H}_2$  environment at 600 °C for 1 hour.

### A Model Framework for the Exsolution Process

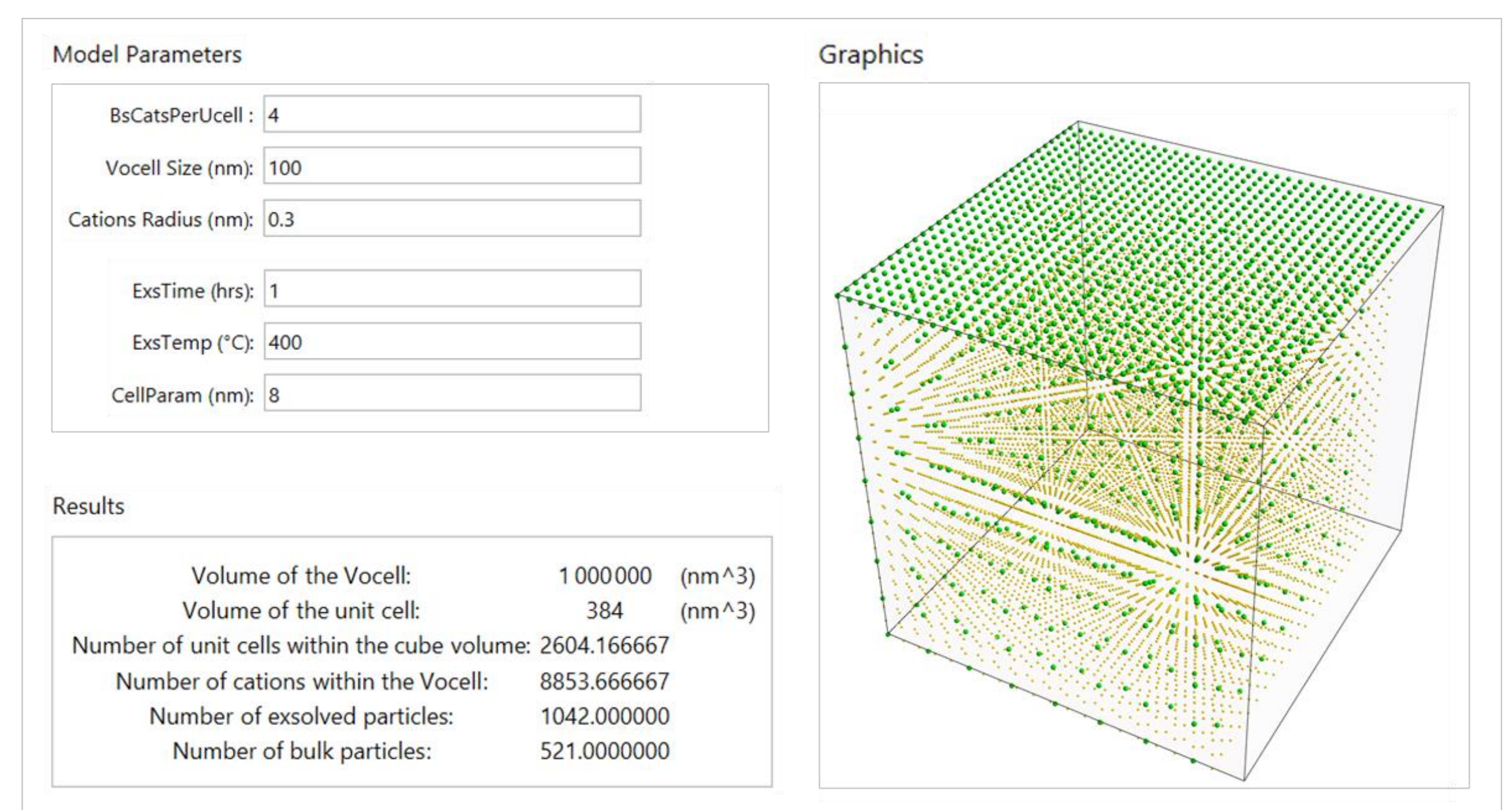


Figure 7: Schematics of a predictive model for the exsolution process for properties tuning in the synthesized perovskite materials.

## CONCLUSION

- ❖ Achieving exsolution at 400 - 600 °C and 1 hour in the novel perovskites has shown the possibility of low temperature operation and enhanced durability of RSOCs.
- ❖ Electrodes fabricated from the novel perovskites can overcome the electrochemical requirements of RSOCs and enhance mode switching between electrolytic and fuel cell.
- ❖ The predictive model for exsolution process when completed will facilitate further optimization of the novel perovskites properties for hydrogen production and power generation.

## FUTURE WORK

- ❖ Exsolution analysis and study of bulk exsolution in the materials;
- ❖ Modelling of the exsolution process for performance optimisation;
- ❖ Fabrication of RSOCs electrodes; and
- ❖ Fabrication of RSOCs for testing and benchmarking.

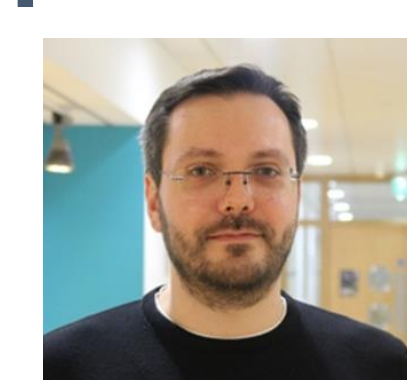
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