

See discussions, stats, and author profiles for this publication at: <https://www.researchgate.net/publication/289756854>

Isolation of some non polar constituents from the fruits of *Faidherbia albida* (Del.) A. Chev.

Article · January 2015

CITATION

1

READS

848

1 author:



Abdullahi Mann

Federal University of Technology Minna

60 PUBLICATIONS 1,110 CITATIONS

SEE PROFILE

Some of the authors of this publication are also working on these related projects:



Phytochemical Investigation and Biological Evaluation of Indigenous Plants for Management of Common Tropical Diseases [View project](#)



Antitrypanosoma and cysteine protease inhibitory activities of selected medicinal plant [View project](#)

**INTERNATIONAL JOURNAL OF ADVANCES IN
PHARMACY, BIOLOGY AND CHEMISTRY**

Research Article

**Isolation of some non polar constituents from the
fruits of *Faidherbia albida* (Del.) A. Chev.**

A.L. Fadipe*, A. Mann, V. Dasaba, D.O. Usman and P. Olajide.

Department of Chemistry, School of Physical Sciences, Federal University of Technology,

Minna, Niger State, Nigeria.

ABSTRACT

Faidherbia albida, also known as *Acacia albida* is a pod-bearing medicinal plant that has been reported to possess several medicinal uses and invariably some bioactives. In this study, the isolation, characterization and identification of some bioactives from the petroleum ether-soluble portion of the methanol extract of the fruits was carried out. This was achieved using techniques, such as extraction, partitioning, fractionation, chromatography, separation and purification to afford five major partially purified fractions. GC-MS of two of these fractions revealed the presence of some saturated and unsaturated fatty acids and fatty acid esters. Further purification of the two fractions led to the isolation of four non-polar compounds that were characterized based on physical, chemical, spectral and literature search. The compounds were identified as stearic acid, oleic acid, linoleic acid, methyl ester and linolenic acid, methyl ester, all reported for the first time from the fruits of *F. albida*.

Keywords: *Faidherbia albida*, non-polar compounds, spectroscopy, fatty acids and fatty acid esters

INTRODUCTION

Medicinal plants over the decades have been shown to have a scientific basis for their use in primary health care in the prevention and treatment of various diseases by several scientists. Such therapeutic uses of such plants have been proved to be as a result of the presence of various anti-nutritive factors, also known as bioactives or phytoconstituents, which act individually or synergistically in such plants. Therefore, the continuous search for such secondary metabolites in plants cannot be undermined. A medicinal plant, such as *Faidherbia albida*, also known as *Acacia albida* of the Mimosaceae/Leguminosae/ Fabaceae (pod-bearing) family; commonly known as Apple ring- Acacia (English), Oju Ologbo (Yoruba), Gawo (Hausa) and Anya-nwona (Igbo) is a thorny, deciduous tropical African tree with deep penetrating root. The leaves are bi-pinnate, bluish -green in color with straight whitish thorns, while, the bark is grey, rough and scaly, when old. It possesses pale cream-colored, long-spiked flowers and has fruits that are twisted, glabrous, and shiny, ranging in color from orange to brown. The fruits are soft and fleshy with narrow

wings that extend about three quarters way round the body with a corky and knobby flesh surrounding a hard woody shell¹. Within the fruits are pod-bearing seeds which are orange-brown in color, twisted in shape, containing almost 10-30 dark, shiny brown ovoid seeds and a tough seed coat; usually eaten by animals^{2, 3, 4}. Traditionally, the plant is used in the treatment of diarrhea^{5, 6}, skin diseases and asthma⁷. The plant is also useful as an anti-inflammatory, anti-haemorrhagic and ophthalmic agent⁵. Locally, the seeds of the plant are eaten by humans as foods during famine^{5, 8}, while powdered pods and seeds are widely used to poison fish in pools⁹. Biologically, the plant's usefulness as anti-microbial¹⁰, anti-diarrhetic, anti-pyretic, anti-inflammatory¹¹, anti-trypanosomiasis¹², anti-diabetic¹³, anti-malarial¹⁴, anti-fungal¹⁵ and nematicidal agent¹⁶ has been reported. Agriculturally, the plant is a nitrogen fixer¹⁷ and has reportedly been used to improve fertility and hence, increase crop yield of crops planted around it¹⁸. Phytochemically, the presence of saponins, tannins, alkaloids, flavonoids, phytates, oxalates, carbohydrates and glycosides has been

reported in the plant^{11,13, 19-21}. Extraction and characterization of the seed oil of the plant revealed that it is quite rich in saturated and unsaturated fatty acids, such as myristic, palmitic, stearic, palmitoleic, oleic and linoleic acid²². GC-FID and GC-MS analysis of the composition of essential oils from the stem bark of the plant revealed the presence of thirty-seven constituents, with α -pinene (a terpene) being the most abundant²³. In view of the absence of much information on the bioactives of the fruits of *F. albida*, the continuous search for secondary metabolites in plants has therefore prompted the extraction, isolation and characterization of more constituents from the fruits of *Faidherbia albida*

MATERIALS AND METHODS

Collection and Identification of Plant Material

Fruits of *Faidherbia albida* were collected from Madari, Jigawa Local Government Council of Kano State, Nigeria; in the month of December, 2014. Identification and authentication was carried out by Dr. (Mrs.) Jemilat Ibrahim of the Department of Medicinal Plant Research and Development (MPR&TM) of National Institute for Pharmaceutical Research and Development, Idu (NIPRD) and a voucher deposited (NIPRD/H/6151).

Extraction of Plant Material

Air-dried powdered fruits of *Faidherbia albida* (700 g) were extracted exhaustively by macerating with 80% methanol. The filtrate was decanted, filtered and concentrated *in vacuo*. Dried extract was weighed and coded crude methanolic extract of fruits of *F. albida* (M).

Partitioning of Crude Extract (M)

The extract, M, was suspended in 500 ml of distilled water, shaken vigorously and set aside for 2 h. The homogenous reddish-brown mixture obtained was then partitioned with petroleum ether (60-80°C, 7x100 ml) and the resulting mixture concentrated *in vacuo*, and evaporated to dryness to obtain an extract coded petroleum ether-soluble portion of methanol extract of *A. albida* (Mp).

Fractionation of Petroleum ether-soluble Portion (Mp)

Two grams of portion Mp was solubilized in petroleum ether (little quantity) and adsorbed onto silica gel (little quantity). It was evaporated to dryness and the homogeneous mixture subjected to fractionation by flash column chromatography using silica gel (mesh 230-400, 60 g) and petroleum ether (60-80°C) by the slurry method. Elution was carried out with varying proportions of petroleum ether: CHCl₃ and eluates monitored by TLC (pre coated). Spots on chromatograms were detected using

sunlight, UV light and iodine crystals. Identical fractions were pooled and concentrated *in vacuo* to afford 5 major fractions, of which fractions obtained from solvent systems petroleum ether: CHCl₃ (4:1) and petroleum ether: CHCl₃ (1:1) were the most promising, with well resolved spots on TLC. These two fractions were evaporated to dryness and coded Mp-1 and Mp-2 respectively.

GC-MS of Fractions Mp-1 and Mp-2

The identification of the compounds present in both fractions Mp-1 and Mp-2 using gas chromatography coupled with mass spectroscopy (GC-MS) was based on direct comparison of the retention times and mass spectral data with those for standard compounds and by computer matching with the Wiley 229, NIST 107 and NIST 21 libraries as well as by comparison of the fragmentation patterns of the mass spectra with those reported in the literature^{24, 25}. Major constituents present in both fractions are presented in Tables 1 and 2 respectively.

Isolation of Compounds Mp-1a and Mp-1b

TLC of fraction Mp-1 using petroleum ether: EtOAc (9:1) as the mobile phase revealed two major and a minor spot. These were resolved by subjecting the fraction (240 mg) to further purification using flash chromatography (silica gel, mesh 230-400, 30 g). Elution was carried out with varying proportions of petroleum ether and EtOAc. Sub-fraction from solvent mixture petroleum ether: EtOAc (19:1) yielded two major spots that were resolved by further purification using preparative thin layer chromatography (1.00mm, petroleum ether: CHCl₃, 9: 1). This gave rise to 2 major bands that were scrapped and triturated with acetone, individually. Concentration of band 1 *in-vacuo* afforded a waxy mixture, which on further purification using PTLC (petroleum ether: CHCl₃, 4: 1) afforded a single spotted compound, coded Mp-1a. The concentration of band 2 *in-vacuo* yielded an oily substance with some minor impurities at the origin. The oil was washed severally with methanol, yielding a methanol-insoluble fraction, which on TLC afforded a single spotted compound, coded Mp-1b.

Isolation of Compounds Mp-2a and Mp-2b

TLC of fraction Mp-2 using petroleum ether: EtOAc (4:1) as the mobile phase revealed five major spots which were resolved by subjecting the fraction (225 mg) to further purification using flash chromatography (silica gel, mesh 230-400, 30g). Elution was carried out with varying proportions of petroleum ether and EtOAc. Sub-fraction (114 mg) from solvent mixture petroleum ether: EtOAc (9:1)

was further purified over column (silica gel, mesh 230-400, 30 g, hexane: CHCl₃) to afford 4 major sub-fractions. Further purification of sub-fraction from hexane: CHCl₃; 3:7 (83 mg) using column (silica gel, mesh 230-400, 30 g, hexane: EtOAc) afforded a single spotted compound each, coded Mp-2a and Mp-2b from solvent systems hexane: ethyl acetate (9:1) and hexane: ethyl acetate (6:4) respectively.

Characterization of Isolated Compounds

Spots on chromatograms were detected using sunlight, UV light and iodine crystals. Chromogenic reagents such as: (i) vanillin-sulphuric acid, heated at 105°C, and (ii) 5% alc. KMnO₄ were also used as spot detectors. UV and IR were both recorded in CHCl₃ using UV spectra (T60 UV-Visible spectrophotometer) and FTIR (Spectrolab MB 3000) respectively; while GC-MS was recorded using GCMS-QP 2010 plus Shimadzu. ¹H-NMR, ¹³C-NMR (proton-decoupled) and DEPT-135 spectra were taken in CDCl₃ on JEOL LA spectrometer operating at 500 MHz. Boiling point was determined using the Siwoloboff's method ²⁶, while melting points were uncorrected.

RESULTS AND DISCUSSION

GC-MS of partially purified fractions Mp-1 and Mp-2 revealed that both fractions were made up of fatty acids and fatty acid esters as shown in both Tables 1 and 2. It was shown that the monounsaturated fatty acid, MUFA (oleic acid) was of higher content in fraction Mp-1 than the saturated fatty acids (stearic acid and palmitic acid), while; only a trace amount of a saturated fatty acid methyl ester, FAME (stearic acid methyl ester) was present in the fraction (Table 1). The spectrum of fraction Mp-2 revealed the presence of only fatty acid methyl esters, FAMES, of which three were unsaturated (linoleic-, linolenic- and oleic acid methyl ester), while; two were saturated (palmitic and eicosanoic acid methyl ester) as shown in Table 2, an indication that both fractions are richer in the unsaturated than saturated fatty acids/fatty acid esters.

Fatty acids (FAs) are long aliphatic/hydrocarbon chain containing a carboxylic acid moiety at one end. They are formed by the base-induced ester hydrolysis of fats and oils (saponification). They exist as both saturated and unsaturated fatty acids, in which the unsaturated could be monounsaturated (MUFA) or poly unsaturated fatty acid (PUFA). They are all a common occurrence in plants and animals ^{27, 28, 29}. Fatty acid methyl esters (FAMES) are also long aliphatic chains, like their fatty acid counterparts, but have their hydrogen group in the COOH moiety being replaced by a -CH₃ group. They are derived by

trans-esterification of fats/oils with methanol ³⁰. They also exist as saturated or unsaturated FAMES.

Structural Elucidation of Isolated Compounds Stearic acid (Compound Mp-1a)

Physical and spectral characterization of compound Mp-1a revealed the following: white waxy solid (17.8 mg), melting point 68.6 -70.4°C (lit: 69.3°C), soluble in hexane and chloroform; slightly soluble in acetone and ethanol; insoluble in water, indicating that it is an apolar organic compound. TLC, single spotted, R_f 0.61 (no color under sunlight; UV active; golden brown in iodine; blue with vanillin-sulphuric acid + heat and deep brown with 5% alc. KMnO₄).

IR (max, cm⁻¹): 3208.21 (OH of an acid), 2857.6 (C-H), 1722.33 (CO of an acid).

UV (max, nm): 199 (saturated COOH).

¹H-NMR (ppm): 10.8 (weak singlet, H-1 of OH), 2.20-2.18 (weak triplet, H-2), 1.53-1.50 (weak multiplet, H-3), 1.27 (sharp singlet, H-4 to H-16), 1.30-1.28 (weak multiplet, H-17) and 0.95-0.92 (weak triplet, H-18).

¹³C-NMR analyzed with DEPT 135 (ppm): revealed nine proton de-coupled peaks at 176.4 (C-1), 34.3 (C-2), 24.1 (C-3), 28.2 (C-4), 28.5 (C-5 and C-15), 28.8 (C-6 to C-14), 30.3 (C-16), 21.9 (C-17) and 14.0 (C-18), of which there are; one quaternary (C-1), sixteen methylene (C-2 to C-17), one methyl (C-18) and no methine groups.

GC-MS (m/z peaks): 284[M⁺, C₁₈H₃₆O₂]⁺, 241[C₁₅H₂₉O₂]⁺, 199[C₁₂H₂₃O₂]⁺, 185 [C₁₁H₂₁O₂]⁺, 171 [C₉H₁₉O₂]⁺, 143 [C₈H₁₅O₂]⁺, 129 [C₇H₁₃O₂]⁺, 115 [C₆H₁₁O₂]⁺, 98[C₇H₁₄]⁺, 85[C₆H₁₃]⁺, 83[C₆H₁₁]⁺, 73[base peak, C₅H₁₃]⁺, 57[C₄H₉]⁺, 55[C₄H₇]⁺, 43[C₃H₆]⁺ and 41[C₃H₅]⁺. GC-MS revealed its molecular formula and weight as C₁₈H₃₆O₂ and 284 gmol⁻¹ respectively.

Compound Mp-1a, was revealed to be a non-polar compound possessing no double bond and has a carboxylic acid functional group at C-1. It was identified as stearic acid (Fig. 1), a common saturated fatty acid (18:0) that has been reported and isolated in several medicinal plants ³¹⁻³⁵.

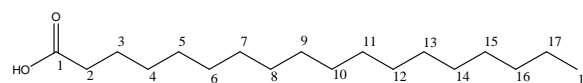


Fig. 1
Structure of Stearic acid /n-Octadecanoic acid/Octanedecane carboxylic acid

Oleic acid (Compound Mp-1b): A yellow viscous oil with a slight odor (15.7 mg), melting point 12.4-14.6°C (lit: 13.5°C), boiling point 358-359°C (lit: 360°C), soluble in benzene, chloroform, dichloromethane and diethyl ether; slightly soluble in acetone, methanol and ethanol; insoluble in water, indicating that it is also an apolar compound. Spot on TLC, R_f 0.52 (no color under sunlight; UV active; golden brown in I_2 crystals; deep blue with vanillin-sulphuric acid, heated at 105°C for 5 minutes and brown with 5% alc. $KMnO_4$).

IR max (cm^{-1}): 3219.1 (OH of an acid), 2993.2 (C-H, aliphatic), 1714.4 (CO of an acid), 1638.7 (C=C).

UV max (nm): 172 (C=C), 202 (unsaturated COOH).

1H -NMR (ppm): 10.7 (weak singlet, H-1); 5.32-5.25 (weak doublet, H-9 and H-10), 2.10-2.04 (weak triplet, H-2), 1.91-1.86 (weak multiplet, H-8 and H-11), 1.52-1.55 (weak multiplet, H-3), 1.34 (slightly sharp singlet, H-7, H-12 and H-17), 1.30 (very sharp singlet, H-4, H-5, H-6, H-13, H-14, H-15 and H-16) and 1.01-0.94 (weak triplet, H-18).

^{13}C -NMR analyzed with DEPT 135 (ppm): revealed thirteen proton-decoupled peaks at 172.7 (C-1), 133.2 (C-9 and C-10), 35.8 (C-2), 30.7 (C-16), 29.8 (C-7 and C-12), 29.3 (C-6, C-13 and C-14), 29.0 (C-5), 28.9 (C-15), 28.7 (C-4), 26.5 (C-8 and C-11), 24.7 (C-3), 21.6 (C-17) and 13.8 (C-18); of which there are: one quaternary (C-1), one methyl (C-18), fourteen methylene (C-2 to C-8 and (C-11 to C-17) and two methine (C-9 and C-10) groups.

GC-MS (m/z): 282 [M^+ , $C_{18}H_{34}O_2$] $^+$, 283 [$M+1$, $C_{18}H_{35}O_2$] $^+$, 264 [$C_{18}H_{33}O$] $^+$, 127 [C_9H_{19}] $^+$, 98 [C_7H_{14}] $^+$, 97 [C_7H_{13}] $^+$, 83 [C_6H_{11}] $^+$, 69 [C_5H_9] $^+$, 55 [base peak, C_4H_7] $^+$, 43 [C_3H_7] $^+$, 41 [C_3H_5] $^+$, 29 [C_2H_5] $^+$ and 27 [C_2H_3] $^+$. Its molecular formula and weight by GC-MS was found to be $C_{18}H_{34}O_2$ and $282g\text{mol}^{-1}$ respectively. The various data obtained for compound Mp-1b revealed that it is a monounsaturated fatty acid (MUFA) possessing a single double bond at positions C-9 and C-10 and a carboxylic acid functional group at C-1. Compound Mp-1b (Fig. 2) identified as oleic acid (18:1, $-9/cis-9$) is the most common monoenoic fatty acid in plants²⁸ and has been reported and isolated in several medicinal plants^{32, 33, 36}.

Linoleic acid, methyl ester (Compound Mp-2a): Physical and spectral characterization of compound Mp-2a revealed the following: Pale yellow oil (16.5mg), melting point -33 to -31.6°C (lit: -35°C), boiling point 207-209°C (lit: 208°C), soluble in

hexane, chloroform and diethyl ether; partially soluble in acetone, methanol and insoluble in both hot and cold water, also an apolar compound. TLC, single spotted, R_f 0.55 (no color under sunlight; UV active; golden brown in iodine; blue with vanillin-sulphuric acid + heat and deep brown with 5% alc. $KMnO_4$).

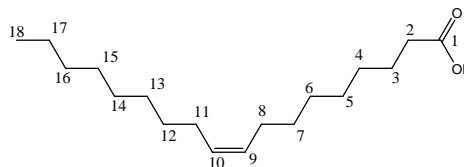


Fig 2
Structure of Oleic acid/cis-Oleic acid/9-Octadecenoic acid (Z)/cis-9-Octadecenoic acid

IR (max, cm^{-1}): 3012.38 (=CH), 2950.1 (-CH₃), 1749.22 (CO of an ester), 1645.33 (C=C) and 1202.62 (O-C-C of an ester).

UV (max, nm): 214 (2x C=C), 211 (unsaturated ester).

1H -NMR (ppm): 5.50-5.42 (weak multiplet, H-9, H-10, H-12 and H-13), 3.70 (sharp singlet, H-1'), 2.65-2.62 (weak triplet, H-11), 2.2-1.98 (weak triplet, H-2), 1.95-1.92 (weak doublet of a triplet, H-8 and H-14), 1.70-1.68 (weak pentet, H-3), 1.40-1.36 (multiplet, H-7, H-15 and H-17), 1.34-1.31 (multiplet, H-4, H-5, H-6 and H-16) and 1.00-0.97 (weak triplet, H-18).

^{13}C -NMR analyzed with DEPT 135 (ppm): revealed sixteen proton de-coupled peaks at 171.6 (C-1), 130.8 (sharp, C-9 and C-13), 125.2 (sharp, C-10 and C-12), 50.7 (C-1'), 31.8 (C-2), 30.4 (C-16), 29.8 (C-7), 29.5 (C-6), 29.4 (C-15), 29.2 (C-5), 28.8 (C-4), 26.7 (sharp, C-8 and C-14), 25.4 (C-11), 25.0 (C-3), 21.9 (C-17) and 13.9 (C-18); of which there are: one quaternary (C-1), two methyl (C-18 and C-1'), twelve methylene (C-2 to C-8, C-11, C-14 to C-17) and four methine (C-9, C-10, C-12 and C-13) groups.

GC-MS (m/z peaks): 294 [M^+ , $C_{19}H_{34}O_2$] $^+$, 263 [$C_{18}H_{31}O_2$] $^+$, 164 [$C_{11}H_{16}O$] $^+$, 150 [$C_{10}H_{14}O$] $^+$, 136 [$C_9H_{12}O$] $^+$, 123 [$C_8H_{10}O$] $^+$, 109 [C_7H_9O] $^+$, 95 [C_7H_{11}] $^+$, 81 [C_6H_9] $^+$, 67 [base peak, C_5H_7] $^+$, 55 [C_4H_7] $^+$ and 41 [C_2H_3] $^+$. GC-MS revealed its molecular formula and weight as $C_{19}H_{34}O_2$ and $294g\text{mol}^{-1}$ respectively. The various peaks obtained for

compound Mp-2a revealed that it is a diene fatty acid methyl ester possessing two double bonds at positions C-9, C-10, C-12 and C-13 and a methyl carboxylate functional group at C-1. Compound Mp-2a (Fig. 3) identified as linoleic acid, methyl ester (18:2, -6/ -9, *cis, cis*-^{9, 12}), an omega-6 polyunsaturated fatty acid, PUFA derivative of linoleic acid, one of the most common dienoic fatty acids, also known as an essential fatty acid²⁸. Linoleic acid, methyl ester has been reported and isolated in several medicinal plants³⁷⁻⁴⁰.

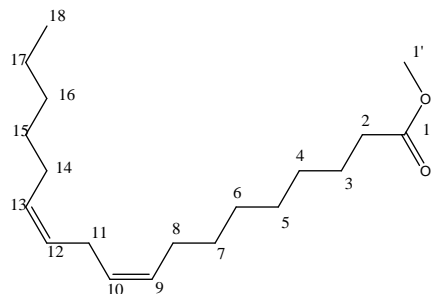


Fig. 3

Structure of Linoleic acid, methyl ester/9, 12-Octadecadienoic acid (Z, Z) methyl ester/Methyl *cis, cis*-9, 12-

**Octadecadienoate/ Methyl linoleate
Linolenic acid, methyl ester (Compound Mp-2b)**

Physical and spectral characterization of compound Mp-2b revealed the following: A colorless viscous oil (18.1mg), melting point -56.2 to -53.1°C (lit: -57 °C), boiling point 180.5-181.7 °C (lit: 182°C), soluble in hexane, chloroform and diethyl ether; partially soluble in acetone, methanol and insoluble in both hot and cold water. TLC, single spotted, R_f 0.58 (no color under sunlight; UV active; golden brown in iodine; blue with vanillin-sulphuric acid + heat and deep brown with 5% alc. $KMnO_4$).

IR (max, cm^{-1}): 3162.38 (=CH), 1743.29 (CO of an ester) 1648.74 (C=C) and 1216.34 (O-C-C of an ester).

UV (max, nm): 221 (3x C=C), 220 (unsaturated ester).

1H -NMR (ppm): 5.46-5.35 (weak multiplet, H-9, H-10, H-12, H-13, H-15 and H-16), 3.67-3.66 (sharp singlet, H-1'), 2.64-2.61 (weak multiplet, H-11 and H-14), 2.29-2.25 (weak triplet, H-2), 2.05-1.99 (weak multiplet, H-17), 1.95-1.91 (weak multiplet, H-8), 1.71-1.67 (weak pentet, H-3), 1.37-1.32 (weak pentet, H-7), 1.30-1.27 (weak multiplet, H-4, H-5 and H-6) and 1.09-1.04 (weak triplet, H-18).

^{13}C -NMR analyzed with DEPT 135 (ppm): revealed seventeen proton de-coupled peaks at 172.8 (C-1), 131.4 (C-9), 129.7 (C-16), 129.5 (sharp, C-12 and C-13), 129.1 (C-15), 128.6 (C-10), 51.2 (C-1'),

34.8 (C-2), 31.1 (C-7), 29.9 (C-6), 29.6 (C-5), 29.0 (C-4), 28.3 (C-8), 26.1 (sharp, C-11 and C-14), 25.7 (C-3), 19.8 (C-17) and 13.9 (C-18); of which there are: one quaternary (C-1), two methyl (C-18 and C-1'), ten methylene (C-2 to C-8, C-11, C-14 and C-17) and six methine (C-9, C-10, C-12, C-13, C-15 and C-16) groups.

GC-MS (m/z peaks): 292[M⁺, C₁₉H₃₂O₂]⁺, 261 [C₁₈H₂₉O]⁺, 149[C₁₀H₁₃O]⁺, 135[C₉H₁₁O]⁺, 121[C₈H₉O]⁺, 108[C₇H₈O]⁺, 93[C₇H₉]⁺, 79[base peak, C₆H₇]⁺, 67[C₅H₇]⁺, 55[C₄H₇]⁺ and 41[C₃H₅]⁺. GC-MS revealed its molecular formula and weight as C₁₉H₃₂O₂ and 292gmol⁻¹ respectively. The spectral data obtained for compound Mp-2b revealed that it is a tri-unsaturated/triene fatty acid methyl ester possessing three double bonds at C-9, C-10, C-12, C-13, C-15 and C-16. It has a methyl carboxylate functional group at C-1, like compound Mp-2a. Compound Mp-2b (Fig. 4) identified as linolenic acid, methyl ester (18:3, -3/ -6/ -9, *cis, cis, cis*-^{9, 12, 15}) is an omega-3 polyunsaturated fatty acid, PUFA derivative of α -linolenic acid, one of the most common trienoic fatty acids, also known as an essential fatty acid²⁸. Linolenic acid, methyl ester has also been reported and isolated in several medicinal plants^{38,40,41}.

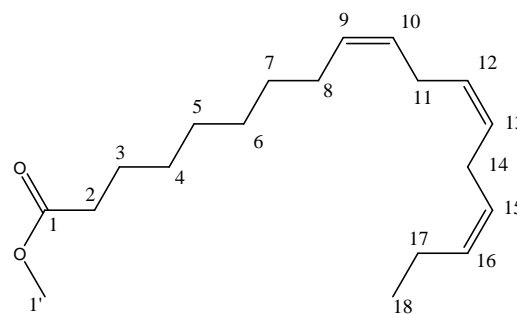


Fig. 4

Structure of Linolenic acid, methyl ester/9, 12, 15-Octadecatrienoic, (Z, Z, Z) methyl ester/ Methyl-all-*cis*-9, 12, 15-octadecatrienoate, Methyl linolenate

The result of this study indicated that the petroleum ether-soluble portion of the methanol extract of the fruits is a rich source of polyunsaturated fatty acids (PUFAs) as well as monounsaturated fatty acids (MUFAs). Moreover, these fatty acids play important role in human growth and development and are positively associated with health and the prevention and treatment of heart disease, arthritis, inflammatory and autoimmune diseases and cancer^{42, 43}. For instance, edible oils which are rich in unsaturated fatty acids especially, monounsaturated fatty acids are known to lower total and LDL cholesterol,

triglycerides, increase HDL cholesterol, help controlling blood pressure, impart antithrombotic properties and help improve insulin sensitivity and thus help in reducing the oxidative stress in the human body^{42, 43}. They are known to reduce the risk of cardiovascular diseases and to have an immunomodulatory effect on conditions such as rheumatoid arthritis^{42, 43}.

CONCLUSION

Fractionation, purification and chromatographic separation of the petroleum ether-soluble portion of the methanol fruit extract of *Faidherbia albida* led to

the isolation, characterization and structural elucidation of stearic acid (saturated free fatty acid, FFA), oleic acid (unsaturated monoenoic fatty acid, MUFA), linoleic acid, methyl ester (unsaturated dienoic fatty acid methyl ester, FAME) and linolenic acid, methyl ester (unsaturated trienoic fatty acid methyl ester, FAME) from the plant. This is the first report of isolation and characterization of these non-polar compounds from the fruits of the plant. Further work will focus on the isolation and characterization of polar secondary metabolites from the plant and the bioassay of such isolates will also be determined.

Table 1
Fatty acids and fatty acid ester identified in fraction Mp-1 of petroleum ether-soluble portion of fruits of *F. albida*

S No	Name of compound	Molecular formula	Molecular weight (gmol ⁻¹)	Base peak	% Abundance
1	Oleic acid	C ₁₈ H ₃₄ O ₂	282	55.05	52.73
2	Stearic acid	C ₁₈ H ₃₆ O ₂	284	73.05	18.93
3	Palmitic acid	C ₁₆ H ₃₂ O ₂	256	88.10	16.88
4	Stearic acid, methyl ester	C ₁₉ H ₃₈ O ₂	298	74.05	3.83

Table 2
Fatty acid esters identified in fraction Mp-2 of petroleum ether-soluble portion of fruits of *F. albida*

S No	Name of compound	Molecular formula	Molecular weight (gmol ⁻¹)	Base peak	% Abundance
1	Linoleic acid, methyl ester	C ₁₉ H ₃₄ O ₂	294	67.05	26.78
2	Linolenic acid, methyl ester	C ₁₉ H ₃₂ O ₂	292	79.10	11.07
3	Palmitic acid, methyl ester	C ₁₇ H ₃₄ O ₂	270	74.05	9.53
4	Oleic acid, methyl ester	C ₁₉ H ₃₆ O ₂	296	55.05	1.50
5	Eicosanoic acid, methyl ester	C ₂₁ H ₄₂ O ₂	326	74.05	1.08

REFERENCES

1. Keay RJ, Onochie CA, Stanfield DP. Nigerian trees. Nigerian National Press Ltd., Apapa, Lagos, 1999, pp. 495.
2. Wood PJ. The botany and distribution of *Faidherbia albida*. In: Vandenbeldt, R. J. (ed.), *Faidherbia albida* in the West African semi-arid tropics. Proceedings of a workshop, 22-26 April, 1991, Niamey, Niger; ICRISAT and ICRAF, Patancheru, A. P. 502324, India, 1992, pp. 206.
3. Coates PM. Trees of Southern Africa, 3rd ed., Stuij, Cape Town, South Africa 2002, pp. 23-24.
4. Orwa C, Mutua A, Kindt R, Jamnadass R, Anthony S. Agroforestry database: a tree reference and selection guide, version 4. 0, World Agro Forestry Centre, Kenya, 2009.
5. Wickens GE, Seif El Din AG, Guinko S, Ibrahim N. Role of *Acacia* species in the rural economy of dry Africa and the near East, FAO-conservation guide, 1995, No. 27.
6. Muhammad S, Mushtaq A and Ashfaq A. Chemistry of the medicinal plants of the genus *Acacia*. Hamdard-medicus, 1998; 4(1): 63 - 67
7. Gill LS. Medicinal uses of trees and plants in Africa. University of Benin Press, Benin, Nigeria, 1999, pp. 27
8. Murunda CT. Use of seed of *Faidherbia albida* (syn. *Acacia albida*) for human consumption during famine periods in the Gokwe communal lands of Zimbabwe, Australian dry zone *Acacias* for human food, CSIRO, Canberra, Australia, 1992: pp. 93-98.
9. Timberlake J, Field guide to the *Acacias* of Zimbabwe, CBC Publishing, Harare, 1990, pp. 160.
10. Kubmarawa D, Ajoku GA, Enwerem NM and Okorie DA. Preliminary phytochemical

- and antimicrobial screening of 50 medicinal plants from Nigeria. *Afr. J. Biotech*, 2007; 6(14): 1690 - 1696.
11. Tijani AY, Uguru MO and Salawu OA. Anti-pyretic, anti-inflammatory and anti-diarrhoeal properties of *Faidherbia albida* in rats. *Afr. J. Biotech*, 2008; 7(6): 696 – 700.
 12. Tijani AY, Uguru MO, Salawu OA, Abubakar A, Onyekwelu NO and Akingbasote JA. Effect of *Faidherbia albida* on biochemical parameters of rats infected with *Trypanosoma brucei brucei*. *Afr. J. Pharm Pharmacol*, 2009; 3(1): 26 – 30.
 13. Salisu Y, Agunu A, Abubakar MS and Ibrahim G. Hypoglycemic effects of *Acacia albida* del. (Mimosaceae) methanolic root bark extract. *Nig. J. Pharm. Sci*, 2009; 8(1): 66 – 67.
 14. Salawu O, Tijani, AB, Nwaeze AA and Agbakwuru A. Antimalarial activity of ethanolic stem bark of *Faidherbia albida* (Del.) a. Chev (Mimosidae) in mice. *Scholars Research Library Archives of Applied Sci. Res*, 2010; 2(5): 261-268.
 15. Bazie S, Ayalew A and Woldetsadik K. Antifungal activity of some plant extracts against *Colletotrichum musae* the cause of post harvest banana Anthracnose. *J. Plant Pathol. & Microbiol*, 2014; 5: 2 – 6.
 16. Umar I, Mamman A. Nematicidal potential of *Faidherbia albida* fruit against *Meloidogyne javanica* on cowpea. *Pak. J. Nematol*, 2014; 32(1): 77-83.
 17. Payne WA, Williams JH, MaiMoussa KA and Stern RD. Crop diversification in the Sahel through use of environmental changes near *Faidherbia albida* (Del) A. chev. *Crop Sci*, 1998; 38(6): 1585 – 1591.
 18. Vandenbeldt RJ. *Faidherbia albida* in the West African semi-arid tropics (Vandenbeldt, R. J. Ed.). Proceedings of a workshop, 22-26 April, 1991, Niamey, Niger, ICRISAT and ICRAF, Patancheru, A. P. 502324, India, 1992; pp. 9 – 17.
 19. Calvalho LH, Brandao MGL, Santos-Filho D, Lopes JLC and Krettli AU. Antimalarial activity of crude extracts from Brazilian plants studied *in-vivo* in *Plasmodium berghei*-infected mice and *in-vitro* against *Plasmodium falciparum* in culture. *Braz. J. Med. and Biol. Res*, 1991; 24(11): 1113 – 1123.
 20. Almahy HA, Nasir OD. Phytochemical and mineral content of the leaves of four Sudanese *Acacia* species. *J. Stored Products and Post and Postharvest Res*, 2011; 2(11): 221 – 226.
 21. Osuntokun OT, Olajubu FA. Comparative study of phytochemical and proximate analysis of seven Nigerian medicinal plants. *Appl. Sci. Res. J*, 2014; 2(1): 10 – 26.
 22. Gunstone FD, Taylor GM, Cornelius JA and Hammonds TW. New tropical seed oils II. Component acids of leguminous and other seed oils. *J Sci. Food and Agric*, 1968; 19(12): 706 – 709.
 23. Ogunbinu AO, Okeniyi S, Flamini G, Cioni PL, Ogunwande IA and Babalola IT. Essential oil composition of *Acacia nilotica* Linn and *Acacia albida* Delile (Leguminosae) from Nigeria. *J. Essential Oil Res*, 2010; 22(6): 540 – 542.
 24. Ruikar A, Torane R, Tambe A, Puranik V and Deshpande N. GC-MS study of a steam volatile matter from *Mimusops elengi*. *Int. J. Chem. Tech. Res*, 2009; 1(2): 158 – 161.
 25. Ogunlesi M, Okiei W and Osibote EA. Analysis of the essential oil from the leaves of *Sesamum radiatum*, a potential medication for male infertility factor, by gas chromatography - mass spectrometry. *Afr. J. Biotech*, 2010; 9(7): 1060 – 1067.
 26. Furniss BS, Hannaford AJ, Smith PWG, Tatchell AR. Vogel's Textbook of Practical Organic Chemistry, 5th edition. Longman Group UK Limited, England, 1989; pp. 242.
 27. Gurr MI, Harwood JL. Fatty acid structure and metabolism, In: Gurr, M. I and Harwood, J. L (eds.), *Lipid Biochemistry: an Introduction*. Chapman and Hall, London, 1991.
 28. Rustan AC, Devon CA (2005). *Encyclopaedia of Life Sciences*. John Wiley and sons Ltd, 2005; pp. 1-7.
 29. Gunstone FD, Harwood JL, Dijkstra AJ. The lipid handbook with cd-rom, 3rd ed. Boca Raton, CRC Press, Florida, 2007.
 30. Vjas AP, Verma JL and Subrahmanyam N. A review of FAME production processes. *Fuel*, 2010; 89(1): 1 – 9.
 31. Keat NB, Umar RU, Lajis NH, Chen TY, Li TY, Rahmani M and Sukari MA. Chemical constituents from weed species of spermacoce (Rubiaceae). *The Malaysian J. Anal. Sci*, 2010; 14(1): 6 – 11.
 32. Palafox J, Navarrete A, Sacramento-Rivero J, Rubio-Atoche C, Escoffie P and Rocha U. Extraction and characterization of oil from *Moringa oleifera* using supercritical CO₂ and traditional solvents. *Am. J. Anal. Chem*, 2012; 3(12A): 946 - 949.
 33. Devi CK, Murugunathan G and Sundaram R. Isolation and chemical constituents from

- Asteracantha longifolia* seeds. Internat. J. Nat. Prod. Res, 2012; 1(1): 7-10.
34. Yadav SK. Isolation and characterization of chemical compounds from flowers of *Cassia alata*, Der Pharma Chemica, 2013; 5(5): 59 – 62.
 35. Ahamed HAU, Uddin MH, Mannan MA, Barua S and Hoque MA. Studies on the isolation, physico-chemical characterization and microbial activities of melon (*Cucumis melo*) seed oil, Internat. J. Innov. & Sci. Res, 2014; 11 (1): 105-111.
 36. Carvalho RHR, Galvao EL, Barros JAC, Conceicao MM and Sousa EMBD. Extraction, fatty acid profile and antioxidant activity of sesame extract (*Sesame indicum* L.). Braz. Chem. Eng, 2012; 29 (2): 409-420.
 37. Chandrasekaran M, Senthilkumar A and Venkatesalu V. Antibacterial and antifungal efficacy of fatty acid methyl esters from leaves of *Sesuvium portulacastrum* L. Eur Rev Med Pharmacol Sci, 2011; 15 (7): 775-780.
 38. Amrouche A, Benmehdi H, Dalile H, Chabane SM, Zaaboub I, Smahi DE, Moussaoui A and Chabane SD. Evaluation of antifungal activity of free fatty acid methyl esters fraction isolated from Algerian *Linum usitatissimum* L. seeds against toxigenic *Aspergillus*, Asian Pac.J. Trop. Biomed, 2013; 3 (6): 443-448.
 39. Borhade S. Synthesis and characterization of fatty acid methyl ester from *Linum usitatissimum* (Linseed) seed oil, World J. Pharm.& Pharmaceut. Sci, 2014; 3 (7): 977-989.
 40. Chabane SD, Amrouche A, Benmedhi H, Malainine H, Chabane SC and Necir NO. Spectral analysis and antifungal activity of fatty acids and fatty acid methyl esters isolated from Algerian *Citrullus colocynthis* L. seeds. Internat. Res. Pharm, 2014; 5(2): 50-56.
 41. Dong M, Oda Y, Hirota M. (10E, 12Z, 15Z)-9- Hydroxy-10, 12, 15-octadecatrienoic acid methyl ester as an anti-inflammatory compound from *Ehretia dicksonii*, Biosci. Biotechnol. Biochem, 2000; 64 (4): 882-886.
 42. Riediger ND, Othman RA, Suh M and Moghadasian MH. A systemic review of the s of n-3 fatty acids in health and disease. J. Am. Dietetic Assoc, 2009; 109(4): 668-679.
 43. Ruxton C, Reed SC, Simpson M and Millington K. The health benefits of omega-3 polyunsaturated fatty acids: a review of the evidence. J. Human Nut. & Dietetics, 2004; 17(5): 449-459.