



**DEPARTMENT OF CHEMISTRY,
SCHOOL OF PHYSICAL SCIENCES,
FEDERAL UNIVERSITY OF TECHNOLOGY,
MINNA**

FIRST SEMESTER EXAMINATION 2021/2022 SESSION

COURSE CODE: CHM 514

UNITS: 2

COURSE TITLE: PHOTOCHEMISTRY AND PERICYCLIC REACTIONS

TIME ALLOWED: 2 HOURS

INSTRUCTIONS: ANSWER ANY THREE (3) QUESTIONS

Q1. 2, 4, 5-Trimethylhexan-3-one can absorb a photon of light and undergo the following photochemical processes at the respective temperatures:

- (i) Norrish Type I cleavage at 60°C (6½ marks)
- (ii) Norrish Type II cleavage at 32°C (7 marks)
- (iii) Intramolecular Hydrogen abstraction at 35°C (6½ marks)

Assuming the molecule phosphoresces to ground state forming sp, sp² and sp³ hybridized molecules;

- a. Provide a suitable mechanism for each process
- b. Name all molecules formed at every stage of each proposed mechanism and identify their type of hybridization.

Q2. (a). β - Carotene, the major phytochemical constituent that accounts for the orange colour in carrots absorbs light at 483 nm. Explain in details the process **(5 marks)**

b. Using 4-Methylpent-4-en-1-yn-3-one as a model:

(i) Indicate the various types of bonding in all atoms of the molecule **(1 mark)**

(ii) Give the distribution of electrons in each molecular orbital **(1 mark)**

(iii) Give a detailed energy diagram of the distribution of all the electrons in their various molecular orbitals from 2 b (ii) above. Assuming the molecule observed both Grothus-Draper and Stark-Einstein laws (only 1 diagram required). **(13 marks)**

Q3. a. List TEN (10) similarities of singlet and triplet states of an organic molecules **(5 marks)**.

b. Predict the type of energy transitions expected in the triplet states of the following molecules:

(i) Diethyl ether (ii) Butanamide (iii) 1-Acetyl-3-hydroxynaphthalene **(6 marks)**

c. With the aid of a Jablonski diagram only, give detailed photophysical processes that a molecule of But-3-en-1-yne will undergo. Considering the transitions that require the lowest energy only. **(DO NOT REPEAT THE SAME PROCESS TWICE) (7 marks)**

d. Rate the photophysical processes in (3c) above in order in which they will occur from the fastest to the slowest.

(2 marks)

Q4. a. List FIVE (5) conditions required for the electrons of a molecule of 2-Methylbut-3-enal to undergo both radiative and non-radiative processes
(5 marks)

b. 2, 3-Dichlorobut-2-ene can undergo photo-isomerization in the presence of propanone.

(i) Suggest a suitable mechanism for the reaction
(5 marks)

(ii) Draw a suitable detailed energy diagram for the entire process
(5 marks)

(iii) What is the importance of addition of propanone?
(5 marks)