DEPARTMENT OF CHEMISTRY



FEDERAL UNIVERSITY OF TECHNOLOGY, MINNA

SECOND SEMESTER EXAMINATION 2021/2022 SESSION

COURSE CODE: CHM326

CREDIT UNIT: 2

COURSE TITLE: INDUSTRIAL CHEMICAL TECHNOLOGY 11

INSTRUCTION: ANSWER ANY THREE (3) QUESTIONS

TIME ALLOWED: 2 HOURS

- Q1a. Write a brief note on the following chemical technological concepts:
 - i. Mass transfer
- ii. Material balance.
- (5 marks each)
- b. Write briefly on automation and remote control of a process stream (5 marks)
- c. Explain the concept 'optimum process condition" as it relate to industrial process marks) (5
- **Q2ai**. Using the thermal condition, explain the various types of reactors. (9 marks)
- ii. Give four (4) basic criteria to be considered when selecting a reactor for the production of a named product. (4
- b. Describe the working principle of fluidized bed catalytic reactor. (5 marks)
- c. Distinguish between homogeneous and heterogenous catalytic process. (2 marks)
- Q3ai. When is a chemical transformation said to occur in kinetic and diffusion region? State how such transformations can be increased in a gas-liquid reaction system. (6 marks)
- b. Explain the various elementary stages involved in a heterogeneous system. (6 marks)
 - c i. Distinguish between heat and energy (2 marks)
 - ii. State the basic laws that governs heat transfer (3 marks)
 - iii. Describe the various mode of heat transfer (3 marks)

Q4a. Enumerate the various steps to be adopted in solving a material balance equation.(6 marks)

b. Justify why continuous and automated operating conditions should be adopted in an manufacture of a given product by a manufacturing outfit. (4 marks)

c. In a chemical technological process, 500kg of 20% by mass of HNO₃ solution was concentrated to 80% in a continuous evaporator. Calculate the production rate of the concentrated solution and the rate of water removal from the evaporator. (10 marks)